

A Rapid Entry to Carbocycles from Carbohydrates *via* Intramolecular Nitronc Cycloaddition

Tony K. M. Shing,* David A. Elsley, and John G. Gillhouley

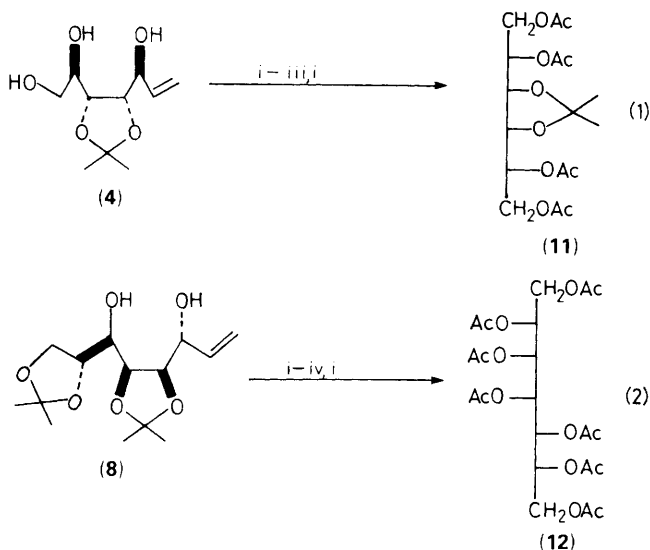
Department of Chemistry, The Victoria University of Manchester, Manchester M13 9PL, U.K.

Cyclopentane (**1**) and cyclohexane (**2**) have been synthesised from D-ribose and D-mannose respectively involving a stereoselective intramolecular nitronc cycloaddition as a key step.

In connection with a programme on the synthesis of polyhydroxylated cyclopentanoid and cyclohexanoid natural products from sugars, *e.g.* aristeromycin¹ and crotepoxide,² we sought to devise a general synthesis of highly oxygenated carbocyclic synthons suitable for further elaboration into a variety of target molecules. Encouraged by a previous report

that hex-5-enoses cyclised to form pentano-isoxazolidines on treatment with *N*-alkylhydroxylamines,³ we now describe a short and flexible method of preparing five- and six-membered carbocycles, cyclopentane (**1**) and cyclohexane (**2**), *via* an intramolecular nitronc cycloaddition (IMNC).

The route to the functionalised cyclopentane (**1**) is shown in



Reagents: i, Ac₂O, pyridine; ii, OsO₄, NaIO₄, aq. dioxan; iii, NaBH₄, aq. EtOH; iv, 1 M HCl, THF.

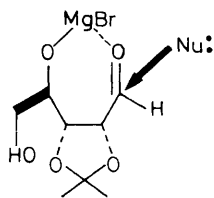


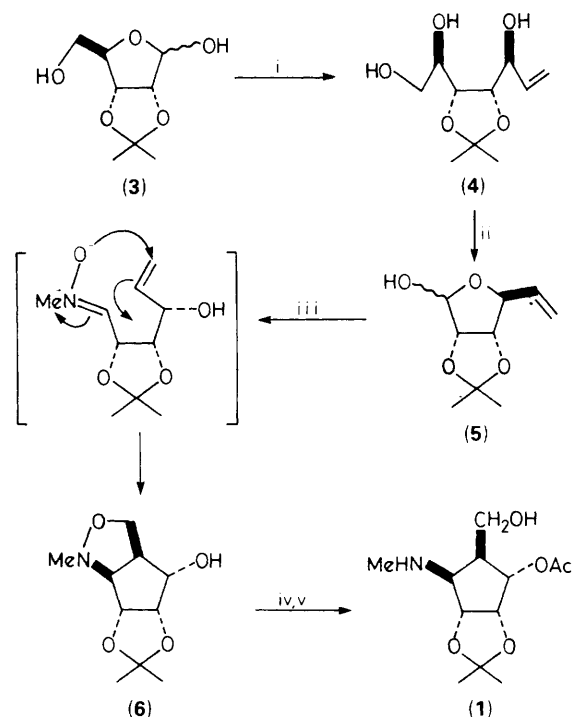
Figure 1

Scheme 1. The acetonide (3),⁴ readily available from D-ribose, reacted with an excess of vinylmagnesium bromide in tetrahydrofuran (THF), presumably *via* a chelation controlled transition model shown in Figure 1, to give triol (4)[†] in *ca.* 9 : 1 diastereoselectivity, m.p. 74 °C; [α]_D -31° (*c* 1.8, CHCl₃). The stereochemistry of the newly formed alcohol in (4) was assigned by analogy⁵ and was confirmed by degradation studies.[‡] Glycol cleavage oxidation of the vicinal diol moiety in (4) afforded the lactol (5), which on heating with *N*-methyl hydroxylamine underwent an IMNC⁶ reaction to give the isoxazolidine (6) as a single diastereoisomer in 94% yield, m.p. 79–80 °C; [α]_D -59° (*c* 0.6, CHCl₃). The stereochemistry of the ring junction in (6) was confirmed by nuclear Overhauser enhancement (n.O.e.) experiments. It is noteworthy that the cycloaddition could proceed with a lactol function and with an unprotected hydroxy group. Acetylation of (6) followed by catalytic hydrogenolysis of the N–O bond then provided the differentially protected cyclopentane (1), [α]_D -84° (*c* 0.9, CHCl₃).

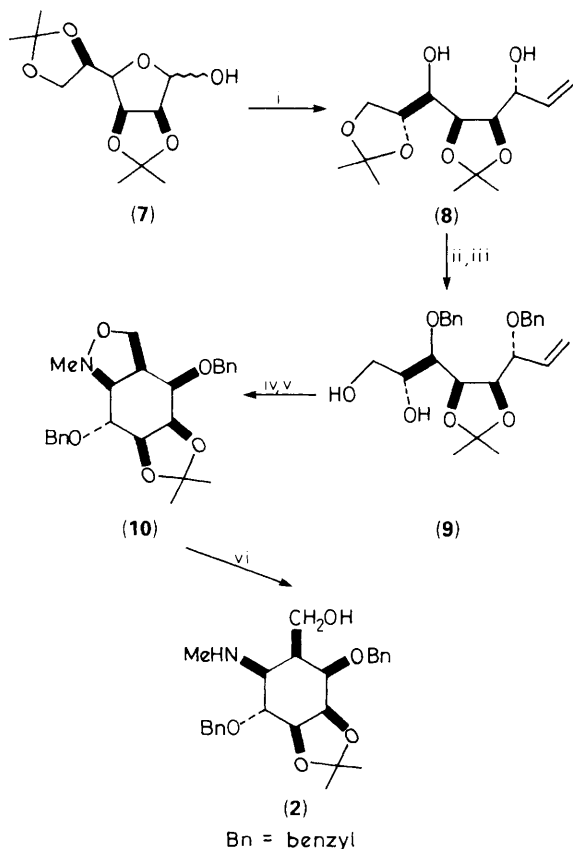
On the other hand, the synthesis of the functionalised cyclohexane (2) is illustrated in Scheme 2. Thus chelation controlled addition of vinylmagnesium bromide to the acetonide (7)⁷ derived from D-mannose gave the diol (8) in *ca.* 5 : 1 diastereoselectivity and in 93% yield, m.p. 54–55 °C; [α]_D +42° (*c* 1.1, CHCl₃). The stereochemistry of the new chiral centre in (8) was confirmed by degradation studies.[‡] The diol

[†] All new compounds gave satisfactory analytical and spectral data.

[‡] The triol (4) was converted into compound (11) which possesses a plane of symmetry, thereby confirming the stereochemistry of the newly formed alcohol [equation (1)]. Transformation of (8) into the known volemitol hepta-acetate⁸ (12) confirmed its stereochemistry [equation (2)].



Scheme 1. Reagents: i, CH₂CHMgBr, THF, (72%); ii, NaIO₄, aq. MeOH, (90%); iii, Me(H)NOH·HCl, NaHCO₃, aq. EtOH, reflux, (94%); iv, Ac₂O, pyridine (85%); v, Pd(OH)₂, H₂, EtOH/AcOH, (75%). Ac = CH₃CO.



Scheme 2. Reagents: i, CH₂CHMgBr, THF, (93%); ii, PhCH₂Br, NaH, THF, (79%); iii, aq. AcOH, (64%); iv, NaIO₄, aq. MeOH; v, Me(H)NOH·HCl, NaHCO₃, aq. ethanol, reflux, [65% from (9)]; vi, Pd(OH)₂, H₂, EtOH/AcOH, (60%).

(8) was protected as the corresponding benzyl ether which was then selectively hydrolysed to form the vicinal diol (9), $[\alpha]_{\text{D}} -120^{\circ}$ (c 1.0, CHCl_3). Glycol cleavage oxidation of (9), followed by immediate reaction with *N*-methyl hydroxylamine, afforded the isoxazolidine (10) as the major adduct in ca. 6:1 stereoselectivity, m.p. 103–104 °C; $[\alpha]_{\text{D}} +11^{\circ}$ (c 1.1, CHCl_3). The stereochemistry of (10) was assigned tentatively by ^1H n.m.r. spectroscopic studies which indicated that the compound existed in a boat conformation in solution. Selective hydrogenolysis of the N–O bond in (10) then yielded the functionalised cyclohexane (2) as an oil, $[\alpha]_{\text{D}} -7^{\circ}$ (c 1.0, CHCl_3).

We thank Professor J. K. Sutherland for discussion and the S.E.R.C. for financial support.

Received, 24th May 1989; Com. 9/02198A

References

- 1 T. Kishi, M. Muroi, T. Kusaka, M. Nishikawa, K. Kamiya, and K. Mizuno, *Chem. Commun.*, 1967, 852.
- 2 For a review on naturally occurring cyclohexene oxides, see C. Thebtaranonth and Y. Thebtaranonth, *Acc. Chem. Res.*, 1986, **19**, 84.
- 3 B. Bernet and A. Vasella, *Helv. Chim. Acta*, 1979, **62**, 1990.
- 4 P. A. Levene and R. S. Tipson, *J. Biol. Chem.*, 1936, **115**, 731.
- 5 R. E. Dolle and K. C. Nicolau, *J. Chem. Soc., Chem. Commun.*, 1985, 1016; J. G. Buchanan, A. D. Dunn, and A. R. Edgar, *Carbohydr. Res.*, 1974, **36**, C5.
- 6 G. Desimoni, G. Tacconi, A. Barco, and G. P. Pollini, 'Natural Products Synthesis Through Pericyclic Reactions,' ACS Monograph 180, American Chemical Society, U.S.A., 1983.
- 7 D. J. Bell, *J. Chem. Soc.*, 1947, 1461.
- 8 W. D. Maclay, R. M. Hann, and C. S. Hudson, *J. Org. Chem.*, 1944, **9**, 293.